## Nasimaluns A and B: neo-Clerodane Diterpenoids from Barringtonia racemosa

Choudhury M. Hasan, Shimul Khan, A. Jabbar, and Mohammad A. Rashid\*

Phytochemical Research Laboratory, Faculty of Pharmacy, University of Dhaka, Dhaka-1000, Bangladesh

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An ethanolic extract of the roots of *Barringtonia racemosa* afforded two novel *neo*-clerodane-type diterpenoids, methyl-15,16-epoxy-12-oxo-3,13(16),14-*neo*-clerodatrien-18,19-olide-17-carboxylate (nasi-malun A, **1**) and dimethyl-15,16-epoxy-3,13(16),14-*neo*-clerodatrien-17,18-dicarboxylate (17-carboxy-methylhardwickiic acid methyl ester, nasimalun B, **2**) by NMR and MS analyses and by comparison of their spectral data with related compunds. The relative stereochemistry of the asymmetric centers in **1** and **2** was determined by selective 1D NOESY experiments.

*Barringtonia* is an important Old World genus of the family Lecythidaceae and consists of 20 species distributed from tropical Africa to Formosa, Polynesia, and northern Australia.<sup>1</sup> *B. racemosa* Blume (Bengali name, "Mohasomudra") is an evergreen tree that grows in the Sunderban of Bangladesh, Sri Lanka, and the west coast of India.<sup>2</sup> Several species found in India are used as folk medicines. The seeds are aromatic and useful in colic and opthalmia.<sup>3</sup> *Barringtonia* species have been shown to contain polyhydroxylated triterpenoids<sup>1,4</sup> and saponins.<sup>5–7</sup> We have examined *B. recemosa*, and in this paper we report the structures of two new *neo*-clerodane diterpenes (**1**, **2**).

The concentrated ethanolic extract of *B. racemosa* was diluted with H<sub>2</sub>O and then extracted with CHCl<sub>3</sub>. Fractionation of the CHCl3-soluble materials by vacuum liquid chromatography (VLC) and TLC afforded two neo-clerodane diterpenes (1, 2). HRFABMS of compound 1 established its molecular formula as C21H24O6. Its IR spectrum displayed bands indicating a furan ring (v 1562, 1507, 873  $cm^{-1}$ ) and three carbonyl groups ( $\nu$  1770, 1730, 1669  $cm^{-1}$ ). The <sup>1</sup>H and <sup>13</sup>C NMR spectra of **1** closely resembled those of tambalin (3), an insect antifeedant *neo*-clerodane isolated from *Tanacetum balsamita*,<sup>8</sup> and bacchasmacranone (4) obtained from Baccharis macraei.9 Typical low-field signals in the <sup>1</sup>H NMR spectrum of **1** at  $\delta$  6.73 (H-14), 7.42 (H-15), and 8.01 (H-16) suggested the presence of a 3-substituted furan ring. The downfield resonance of H-16 revealed that the furan ring was conjugated with a carbonyl group,<sup>10</sup> which was confirmed by the HMBC correlation of H-14 with the ketone carbon at  $\delta_{\rm C}$  193.6 (C-12). This structural fragment was identical to that found in 15,16-epoxy-12oxo-8(17),13(16),14-labdatrien-20,19-olide.<sup>11</sup> The <sup>1</sup>H NMR spectrum of **1** further showed signals for an  $\alpha$ . $\beta$ -unsaturated 18,19-clerodanolide at  $\delta$  6.74 (H-3), 3.93 (H-19a), and 4.33 (H-19b).<sup>9</sup> HMBC correlations from H-3 to C-5 and C-18 and from H-19b to C-4, C-5, C-6, and C-18 confirmed the 18,19-olide functionality in 1. The AB quartet centered at  $\delta$  2.83 and 3.04 assignable to H<sub>2</sub>-11 demonstrated connectivities over two bonds to  $\delta_{\rm C}$  39.6 (C-9) and C-12 and over three bonds to  $\delta_{\rm C}$  48.7 (C-8), 46.7 (C-10), and 19.2 (C-20) in the HMBC spectrum. The last correlation placed the single methyl group resonating at  $\delta$  0.82 on C-9. The doublet of doublets at  $\delta$  3.21 was attributed to H-8, and the HMBC correlation from this proton to the ester carbonyl at  $\delta$  174.0 placed the carboxymethyl substituent  $(\delta_{\rm H} \ 3.60; \ \delta_{\rm C} \ 51.4)$  on C-8.

It was possible to trace all of the proton-proton spin systems in **1** with data from a COSY-45 experiment. Heteronuclear correlation experiments (HMBC and HMQC) allowed unambiguous assignment of all <sup>1</sup>H and <sup>13</sup>C NMR resonances in 1. The relative stereochemistry of the chiral centers in 1 was determined by selective 1D NOESY experiments. Irradiation at the resonance frequency of H-10 produced significant NOESY correlations with H-1eq, H-6ax, and H-8, while H<sub>3</sub>-20 showed strong NOESY interactions with H-1ax, H-7ax, H<sub>2</sub>-19, and both of the C-11 methylene proton resonances. This established a cis relationship between H-8 and H-10 and between the methyl group at C-9 and  $H_2$ -19. Thus, the side chain at C-9 was assigned the equatorial configuration. The key NOE correlations are depicted on the structure 1. On this basis, the new diterpene was identified as methyl-15,16epoxy-12-oxo-3,13(16),14-neo-clerodatrien-18,19-olide-17carboxylate, for which we have proposed the trivial name nasimalun A (1).

The molecular formula for 2 was established as  $C_{22}H_{30}O_5$ from HRFABMS. The <sup>13</sup>C NMR spectrum showed 22 signals, while DEPT and HMQC experiments confirmed that 16 out of the 22 carbons in 2 were attached to protons. The <sup>1</sup>H and <sup>13</sup>C NMR spectral data of 2 were, in part, identical to those reported for hardwickiic acid methyl ester (5),<sup>12</sup> suggesting a close structural similarity between these two compounds. However, resonances appropriate for a secondary methyl group in 5 were absent in the spectra of 2 and were replaced with resonances indicative of a carboxymethyl residue. In fact, the <sup>1</sup>H and <sup>13</sup>C NMR spectra of **2** displayed signals for two carboxymethyl groups, as compared to one in case of hardwickiic acid methyl ester (5). The second carboxymethyl moiety in 2 was placed at C-8 as determined by HMBC experiments. The relative stereochemistry of compound 2 was also established by 1D NOESY experiments as shown. NOESY data established the cis relationship between H-8 and H-10 and between H<sub>3</sub>-19 and H<sub>3</sub>-20. Thus, compound 2 was identified as dimethyl-15,16-epoxy-3,13(16),14-neo-clerodatrien-17,18-dicarboxylate (17-carboxymethylhardwickiic acid methyl ester, **2**) for which the trivial name nasimalun B has been proposed.

## **Experimental Section**

**General Experimental Procedures.** Optical rotations were measured on a JASCO DIP-370 polarimeter using a sodium lamp (589 nm). UV and IR spectra were obtained on V-500 UV/VIS (JASCO) and IR-230 (JASCO) spectrophotometers, respectively. <sup>1</sup>H NMR spectra were obtained in CDCl<sub>3</sub> on a FG 2Hx54 T=25 A600 instrument operating at 600 MHz,

<sup>\*</sup> To whom correspondence should be addressed. Present address: SAIC Frederick, NCI-Frederick Cancer Research and Development Center, Bldg. 560, Rm. 32-63B, Post Box B, Frederick, MD 21702. Tel.: (301) 846-1295. Fax: (301) 846-6157. E-mail: rashid@mail.ncifcrf.gov.



Arrows indicate key NOESY interactions in 1 and 2.

while the <sup>13</sup>C NMR spectra were obtained in the same instrument at 150 MHz using TMS as internal standard. The chemical shifts ( $\delta$ ) and coupling constants (J) are expressed in parts per million (ppm) and hertz (Hz), respectively. Inversedetected heteronuclear correlations were measured using the HMQC (optimized for  ${}^{1}J_{CH} = 145$  Hz) and HMBC (optimized for  ${}^{n}J_{CH} = 8.3$  Hz) pulse sequences with a pulsed-field gradient. 1D NOESY experiments were carried out with the gNOESY pulse sequence using the excitation bandwidth of 20 Hz. EIMS and FABMS were recorded on Hitachi-U and Harata-013 spectrometers, respectively.

Plant Material. Roots of B. racemosa were collected from Khulna, Bangladesh, in November 1998. The sample was identified by Mr. Manzurul Quader Mian, Bangladesh National Herbarium, Dhaka, where a voucher specimen has been deposited (DACB accession no. 28 062).

Extraction and Isolation. Air-dried roots (2.0 kg) were cut into small pieces and extracted with 95% EtOH at room temperature. Evaporation of solvent under reduced pressure gave a semisolid mass (30.0 g). A portion (4.0 gm) of this residue was diluted with H<sub>2</sub>O and extracted with CHCl<sub>3</sub> (200 mL  $\times$  3). The CHCl<sub>3</sub>-soluble material was subjected to fractionation by VLC over Si gel (60 H) using petroleum ether (60-80 °C), EtOAc, and MeOH in order of increasing polarity. The eluates were combined on the basis of TLC analysis to provide a total of 18 fractions. Evaporation of solvents from fraction 7, followed by washings of the crystalline deposits with petroleum ether (60-80 °C) and EtOAc mixture yielded 1 (12.0 mg). Repeated preparative TLC of fraction 6 over Si gel  $PF_{254}$ , using toluene-EtOAc (95:5) as the mobile phase, afforded 8.0 mg of compound 2.

**Nasimalun A (1):** white amorphous powder;  $[\alpha]_D - 85^\circ$  (*c* 0.2, CHCl<sub>3</sub>); UV (MeOH)  $\lambda_{max}$  253 (log  $\epsilon$  3.54), 207 (log  $\epsilon$  4.34) nm; IR (film) v<sub>max</sub> 3146, 2946, 1770, 1730, 1669, 1562, 1507, 1460, 1433, 1411, 1394, 1366, 1322, 1277, 1194, 1144, 1118, 1036, 1016, 981, 940, 915, 873, 823, 807, 771, 753 cm<sup>-1</sup>;<sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz)  $\delta$  8.01 (1H, s, H-16), 7.42 (1H, d, J = 2.0 Hz, H-15), 6.74 (1H, dd, J = 8.0, 2.0 Hz, H-3), 6.73 (1H, d, J = 2.0 Hz, H-14), 4.33 (1H, d, J = 8.0 Hz, H-19b), 3.93 (1H, dd, J = 8.0, 2.0 Hz, H-19a), 3.60 (3H, s, COOCH<sub>3</sub>), 3.21 (1H, dd, J = 13.5, 4.5 Hz, H-8ax), 3.04 (1H, d, J = 18.0, H-11b), 2.83 (1H, d, J = 18.0, H-11a), 2.73 (1H, dd, J = 11.0, 2.0 Hz, H-10ax), 2.28 (1H, m, H-2eq), 2.22 (1H, m, H-2ax), 2.03 (1H, dddd, J = 13.5, 13.5, 13.5, 4.5 Hz, H-7ax), 1.98 (1H, ddd, J = 13.5, 4.5, 3.0 Hz, H-6eq), 1.87 (1H, dddd, J = 13.5, 4.5, 4.5,

3.0 Hz, H-7eq), 1.64 (1H, dddd, J = 11.0, 2.0, 2.0, 2.0 Hz, H-1eq), 1.36 (1H, dddd, J = 13.5, 13.5, 4.5, 2.0 Hz, H-6ax), 1.09 (1H, dddd, J = 11.0, 11.0, 11.0, 4.0 Hz, H-1ax), 0.82 (3H, s, H-20); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 150 MHz)  $\delta$  193.6 (s, C-12), 174.0 (s, C-17), 169.0 (s, C-18), 147.1 (d, C-16), 144.3 (d, C-15), 137.8 (s, C-4), 136.2 (d, C-3), 128.6 (s, C-13), 108.5 (d, C-14), 71.4 (t, C-19), 51.4 (q, COOCH<sub>3</sub>), 48.7 (d, C-8), 46.7 (d, C-10), 46.5 (t, C-11), 45.1 (s, C-5), 39.6 (s, C-9), 33.2 (t, C-6), 27.3 (t, C-2), 22.1 (t, C-7), 20.1 (t, C-1), 19.2 (q, C-20); HRFABMS m/z [MH]+ 373.1640 (calcd 373.1651 for C<sub>21</sub>H<sub>25</sub>O<sub>6</sub>).

**Nasimalun B (2):** white gum; [α]<sub>D</sub> –95° (*c* 0.1, CHCl<sub>3</sub>); UV (MeOH)  $\lambda_{\text{max}}$  205 (log  $\epsilon$  4.27) nm; IR (film)  $\nu_{\text{max}}$  2952, 1733, 1706, 1558, 1506, 1456, 1436, 1362, 1252, 1188, 1143, 1067, 1023, 743, 873, 770, 760 cm<sup>-1</sup>;<sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz)  $\delta$  7.36 (1H, dd, J = 2.0, 1.0, H-15), 7.22 (1H, s, H-16), 6.65 (1H, dd, J = 5.0, 3.0 Hz, H-3), 6.28 (1H, d, J = 2.0 Hz, H-14),3.71 (3H, s, COOCH<sub>3</sub>-4), 3.68 (3H, s, COOCH<sub>3</sub>-8), 2.59 (1H, dd, J = 13.5, 4.0 Hz, H-8ax), 2.53 (1H, ddd, J = 13.5, 13.5, 3.5 Hz, H-12a), 2.44 (1H, ddd, J = 13.5, 3.0, 3.0 Hz, H-6eq), 2.33 (1H, ddd, J = 20.0, 5.0, 5.0 Hz, H-2eq), 2.21 (1H, m, H-12b),2.19 (1H, m, H-2ax), 2.05 (1H, dddd, J = 13.5, 13.5, 13.5, 3.0 Hz, H-7ax), 1.77 (1H, ddd, J = 18.0, 13.5, 5.0 Hz, H-11a), 1.70 (1H, dddd, J = 13.5, 5.0, 5.0, 5.0 Hz, H-1eq), 1.66 (1H, m, H-7eq), 1.53 (1H, dddd, J = 13.5, 13.5, 13.5, 5.0 Hz, H-1ax), 1.45 (1H, m, H-11b), 1.43 (1H, dd, J = 12.0, 6.0 Hz, H-10ax), 1.34 (3H, s, H-19), 1.13 (1H, ddd, J = 13.5, 13.5, 4.0 Hz, H-6ax), 0.95 (3H, s, H-20);  $^{13}\mathrm{C}$  NMR (CDCl\_3, 150 MHz)  $\delta$  174.9 (s, C-17), 167.5 (s, C-18), 142.7 (d, C-15), 141.8 (s, C-4), 138.5 (d, C-16), 137.2 (d, C-3), 125.1 (s, C-13), 111.0 (d, C-14), 51.2 (q, COOCH<sub>3</sub>-4), 51.1 (q, COOCH<sub>3</sub>-8), 49.1 (d, C-8), 46.4 (d, C-10), 40.9 (t, C-11), 38.7 (s, C-9), 37.3 (s, C-5), 34.9 (t, C-6), 26.9 (t, C-2), 21.6 (t, C-7), 20.8 (q, C-19), 19.9 (q, C-20), 18.1 (t, C-12), 17.2 (t, C-1); HRFABMS m/z [MH]+ 375.2167 (calcd 375.2171 for C<sub>22</sub>H<sub>31</sub>O<sub>5</sub>).

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## **References and Notes**

- (1) Subra, G. S. R.; Prasanna, S.; Kumar, V. P.; Yadagiri, B. Indian J. *Chem.* **1986**, *25B*, 113–122. Hooker J. D. *The Flora of British India* (Indian repr.; Periodical
- (2)(a) Honker J. D. Horn From Physics India (Indian Tepl., Terroration Experts: Delhi, (Published by L. Reeve & Co. Ltd.: Kent, England, 1879), 1961; Vol. 2, pp 507–508.
  (3) Nadkarni, K. M. Indian Materia Medica (2nd repr. of the 3rd rev.
- and enlarged ed.), 1982; Vol. 1, p 177.
  (4) Anantaraman, R.; Pillai, K. S. *J. Chem. Soc.* **1956**, 4369–4373.
  (5) Pal, B. C.; Achari, B.; Price, K. R. *Phytochemistry* **1991**, *30*, 4177–
- 4179.
- (6) Guinchard, A.; Massiot, G.; Sevenet, T.; Pusser, J.; Lavaud, C. Book of Abstracts; Luijendik, T., Graff, P. D., Remmelzwaal, A., Verpoorte, R., Eds.; Leiden University: The Netherlands, 1999; p 010.
- Pal, B. C.; Chaudhuri, T.; Yoshikawa, K.; Arihara, S. Phytochemistry (7)1994, 35, 1315-1318.
- Kubo, I.; Jamalamadaka, V.; Kamikawa, T.; Takahashi, K.; Tabata, K.; Kusumi, T. *Chem. Lett.* **1996**, 441–442. (8)
- Gambaro, V.; Chamy, M. C.; Garbarino, J. A.; San-Martin, A.; Castillo, M. Phytochemistry 1986, 25, 2175-3177.
- (10) (a) Bohlman, F.; Ždero, C.; Grenz, M. Chem. Ber. 1977, 110, 1034-1041. (b) Jefferies, P. R.; Payne, T. G.; Raston, C. L.; White, A. H. Aust. J. Chem. **1981**, *34*, 1001–1007. (c) Nagashima, F.; Asakawa, Y. Phytochemistry **1990**, *29*, 3229–3231.
- (11) Qais, N.; Mandal, M. M.; Rashid, M. M.; Jabbar, A.; Koshino, H.; Nagasawa, K.; Nakata, T. *J. Nat. Prod.* **1998**, *61*, 156–157. Costa, M.; Tanaka, C. M. A.; Imamura, P. M.; Marsaioli, A. J.
- (12)Phytochemistry 1999, 50, 117-122.

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